

Derived Oxy-acids and salts Nomenclature

Oxy acids and salts			
Write the chemical formula		Write the Chemical Name	
1	Potassium chlorate	26	$KClO_2$
2	Sodium perchlorate	27	K_2CO_3
3	Sodium carbonate	28	Na_3PO_4
4	Magnesium phosphite	29	Ag_2O
5	Sodium nitrate	30	$Fe(ClO_3)_2$
6	Aluminum sulfite	31	$KClO_3$
7	Ferric sulfate	32	$Ga(BrO_4)_3$
8	Copper (I) permanganate	33	$Pt(ClO)_4$
9	Iron (III) acetate	34	$(NH_4)_2SO_2$
10	Cobalt (III) chlorite	35	$Zn_3(PO_4)_2$
11	Gold (I) phosphate	36	$KClO$
12	Ammonium hypophosphite	37	$KClO_4$
13	Aluminum silicate	38	$FeSO_3$
14	Tin (IV) hypochlorite	39	$(NH_4)_2SO_4$
15	Magnesium chlorate	40	$Al(NO_3)_3$
16	Mercuric perchlorate	41	Li_2CO_3
17	Magnesium phosphate	42	Ag_3PO_4
18	Sodium phosphite	43	$Mg(ClO_4)_2$
19	Ammonium nitrate	44	$NaClO_2$
20	Magnesium sulfite	45	$CaSO_3$
21	Aurous sulfate	46	NH_4NO_2
22	Aluminum nitrite	47	$ZnCO_3$
23	Cuprous chlorite	48	$BaSO_4$
24	Ammonium chlorate	49	Ag_3PO_4
25	Cobalt (I) hyposulfite	50	K_2CrO_2

Nomenclature Worksheet (a mixture of all types)

Write the chemical formula		Write the chemical name	
1	Silver chloride	26	HF
2	Gold (I) sulfite	27	FeCl ₂
3	Sodium sulfate	28	Fe(NO ₃) ₃
4	Mercury (I) oxide	29	CuSO ₄
5	Potassium phosphate	30	SO ₃
6	Manganese (II) chloride	31	Pb(ClO ₄) ₂
7	Lead (II) perchlorate	32	CCl ₄
8	Aluminum hyponitrite	33	Ca(ClO) ₂
9	Iron (II) phosphate	34	ZnSO ₄
10	Copper (II) sulfate	35	PbO ₂
11	Magnesium carbonite	36	SiO ₂
12	Zinc bromide	37	CaCl ₂
13	Antimony (III) sulfide	38	HgI
14	Carbon monoxide	39	P ₂ O ₃
15	Calcium periodate	40	MgSO ₃
16	Arsenic (II) iodide	41	SnCl ₄
17	Tin (II) carbonate	42	(NH ₄) ₂ SO ₃
18	Magnesium chlorate	43	Ba(OH) ₂
19	Silicon dioxide	44	KBrO ₂
20	Ammonium chloride	45	AgNO ₃
21	Zinc carbonite	46	KIO ₂
22	Gold (I) hypophosphite	47	HgCl ₂
23	Calcium hydride	48	Fe(NO ₃) ₂
24	Ammonium sulfate	49	Au ₂ SO ₃
25	Dinitrogen tetroxide	50	K ₂ CrO ₄