## WODSS SCIENCE

Name: \_\_\_\_\_

## SNC 2DI

Date: \_\_\_\_\_

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## Forming lons

Element – Standard Atomic Notation	Bohr diagram of neutral atom (name the atom)	Bohr diagram of ion (name the ion)	Calculation of ion's charge
Positive Ions (from Metals)			
Li			<b>p</b> <sup>+1</sup> =
p <sup>+1</sup> =			+ e <sup>-1</sup> =
e <sup>-1</sup> =			
nº =			
Ве			<b>p</b> <sup>+1</sup> =
p <sup>+1</sup> =			+ e <sup>-1</sup> =
e <sup>-1</sup> =			
nº =			
В			p <sup>+1</sup> =
p <sup>+1</sup> =			+ e <sup>-1</sup> =
e <sup>-1</sup> =			
n <sup>o</sup> =			
Negative Ions (from Non-metals)			
Ν			p <sup>+1</sup> =
p <sup>+1</sup> =			+ e <sup>-1</sup> =
e <sup>-1</sup> =			
nº =			
0			p <sup>+1</sup> =
p <sup>+1</sup> =			+ e <sup>-1</sup> =
e <sup>-1</sup> =			
nº =			
F			p <sup>+1</sup> =
<b>p</b> <sup>+1</sup> =			+ e <sup>-1</sup> =
<b>e</b> <sup>-1</sup> =			
n <sup>o</sup> =			