

WODSS SCIENCE

SNC 2DI

Molecular Compounds - Extra Practice Worksheet

If the name of the compound is given, write the formula. If the formula is given, write the name.

1. CF_4 _____
2. N_2O_5 _____
3. CS_2 _____
4. SO_3 _____
5. P_4O_8 _____
6. iodine tribromide _____
7. chlorine dioxide _____
8. sulfur hexafluoride _____
9. difluorine octachloride _____
10. tribromine nonatelluride _____
11. H_2O _____
12. P_2S_4 _____
13. N_2O_4 _____
14. XeF_4 _____
15. Si_4 _____
16. carbon dioxide _____
17. trinitrogen hexabromide _____
18. diiodine heptaselenide _____
19. CO _____
20. dicarbon octafluoride _____
21. P_4O_{10} _____
22. Si_3N_4 _____
23. Cl_2S_7 _____
24. NBr_5 _____
25. phosphorus trichloride _____
26. PI_3 _____
27. disulfur trioxide _____
28. PCl_5 _____
29. diiodine dichloride _____
30. dinitrogen monoxide _____
31. I_4O_9 _____
32. dihydrogen monoxide _____