

NAMING COMPOUNDS AND WRITING FORMULAS - Extra Practice (4)

1. Write the formula for each of the following compounds:

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|------------------------|-----------------------|----------------------|
| a) sodium fluoride | k) zinc nitride | u) magnesium oxide |
| b) silver nitride | l) strontium oxide | v) silver fluoride |
| c) aluminum chloride | m) aluminum carbide | w) calcium bromide |
| d) barium oxide | n) lithium sulfide | x) magnesium nitride |
| e) magnesium iodide | o) beryllium iodide | y) cesium chloride |
| f) calcium sulfide | p) hydrogen bromide | z) lithium iodide |
| g) lithium oxide | q) potassium chloride | |
| h) barium sulfide | r) silver sulfide | |
| i) potassium phosphide | s) zinc carbide | |
| j) magnesium sulfide | t) calcium nitride | |

2. Name each of the following compounds:

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|--------------------------|----------------------------|----------------------------|
| a) Na_2O | k) Zn_3P_2 | u) Li_2O |
| b) Li_4C | l) Ba_3N_2 | v) BeF |
| c) MgBr_2 | m) MgO | w) Na_3N_2 |
| d) CsI | n) CaS | x) Mg_3P_2 |
| e) Ag_3N | o) BeO | y) Al_2O_3 |
| f) Sr_2C | p) ZnBr_2 | z) KF |
| g) CaCl_2 | q) NaF | |
| h) BaO | r) Sr_3P_2 | |
| i) AlBr_3 | s) AgI | |
| j) H_2S | t) AlN | |