WODSS SCIENCE SCH 4CI

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Date:

Organic Families and Functional Groups

A ______ is a particular combination of atoms that contribute to the physical and chemical properties of a substance. Compounds that have the same functional group and therefore similar properties are grouped together into _____.

Example: Alkenes are an organic family. The functional group in this family is a

Some functional groups are groups of atoms that attach onto hydrocarbon skeleton chains:

- a) Hydroxyl groups (found in alcohols)
- b) Carbonyl groups (found in aldehydes and ketones)
- c) Carboxyl groups (found in carboxylic acids)
- d) Amine groups (found in amines and amides)

Remember: hydrocarbon skeleton chains are _____

Many functional groups are made of ______ bonds, which give the molecules a slightly positive and negative end. This causes the force of attraction between molecules to be stronger, causing the molecules to have higher melting points and boiling points. **Organic Families**

nilies are listed in the table below. We will learn how

The different organic families are listed in the table below. We will learn how to name compounds in the starred families only. The "R" group in the general formula is an alkyl group, which is just the hydrocarbon chain to which the functional group is attached.

	Family Name	General Formula	Name
1.	*Alkanes		
2.	*Alkenes		
3.	*Alkynes		
4.	*Alcohols		
5.	Ethers		
6.	*Aldehydes		
7.	*Ketones		
8.	*Carboxylic Acids		
9.	Esters		

10.	Amines	
11.	Amides	

Ex. 1 Which family do the following molecules belong to?



Alcohols (R-OH)

Naming Alcohols:

- ANE ending \rightarrow _____
- ANE ending → _____
 if the OH group is in the middle, use a ______ to indicate its position (count from the closest end)

Properties:

- Most are _____
- Alcohols are _____ and soluble in water
- Undergo ______ (produce only CO₂ and H₂O)

Examples:



Naming Aldehydes:

- ANE ending→
- Carbonyl group is always on the END of the chain in an aldehyde so no numbering is needed.

Examples:



Naming Ketones:

Examples:



Naming Carboxylic Acids:

- ANE ending→ _____
- ANE ending → ______
 Carboxyl group is always at the ______ of the chain → no numbers needed.

Examples: